Electrochemical Treatments to Significantly Extend the Service Life of Reinforced Concrete Structures

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Outline

- Corrosion basics
- What are electrochemical treatments and how do they work?
- Project case histories
 - I-480 Omaha, Nebraska Electrochemical chloride extraction
 - University of Chicago Re-alkalization







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Corrosion

- Electrochemical reaction
- Requires
 - Moisture
 - Electrolyte concrete
 - Metallic path steel
- Anode
 - Where rust is formed
- Cathode
 - No section loss



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Corrosion of Reinforced Concrete

- Concrete is naturally alkaline
 - pH of about 13
- Steel is naturally passive at this alkalinity
 - Formation of passive layer
- Passive layer can be destroyed by;
 - Chlorides
 - Carbonation









Chloride Induced

- Chloride ions diffuse into concrete and destroy steel's passive layer
- Source of chlorides
 - Marine environments
 - De-icing salts
 - Chemical/processing plants
 - Cast into concrete
- Chlorides are not consumed in corrosion reaction, therefore, once threshold concentration reached, corrosion can occur unabated



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Carbonation

- Carbon dioxide permeates into concrete
- Reduces pH of concrete
 - CO2 reacts with free lime, Ca(OH) 2, resulting in CaCO3 and H2O
- Reduced pH de-passivates steel
- Often seen when
 - Concrete permeability is high
 - Industrial sites
 - Very old structures carbonation is a result of time and exposure







CONCRETE PRESERVATION PROCESS





Electrochemical Chloride Extraction









Electrochemical Chloride Extraction

- Application of temporary impressed current to draw chlorides out of concrete and repassivate the steel.
- Addresses the root cause of chloride induced corrosion
- Chloride levels in concrete are significantly reduced by ECE
- Alkalinity is increased at the level of the steel
 - Increases the chloride concentration required to reinitiate corrosion
- Reinforcing steel is returned to a passive, non-corroding state
 - For as long as chlorides can be kept from the steel

































I-480 Omaha • Rehabilitation to I—480 included deck replacement and substructure rehabilitation

• ECE completed on hammer head piers in 2002

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Re-Alkalization



































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University of Chicago University Hall Façade Repair

- Corrosion deterioration occurring due to carbonation of the concrete
- Re-alkalization conducted to façade
 - 72,300 ft² of concrete surface area
 - Completed spring of 2018







Prior to Treatment





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Temporary Anode





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Connection to Reinforcement





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Post Treatment





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Post Treatment









Sustainability I-480

- 6,700 yd³ of concrete preserved prevents
 - 6,530 pounds of nitrous oxides
 - 3,663 tons of carbon dioxide
 - Equivalent to annual emissions of about 833 people.
 - 13,191 tons of natural resources
 - Potable water to fulfill the daily needs of 1,695 people,
 - Waste generation 15,975 tons
 - Heat to boil 20 Olympic-sized swimming pools.

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Sustainability University of Chicago

- 1,300 yd³ of concrete preserved prevents
 - 1,267 pounds of nitrous oxides
 - 711 tons of carbon dioxide
 - Equivalent to annual emissions of about 162 people.
 - 2,559 tons of natural resources
 - Potable water to fulfill the daily needs of 329 people,
 - Waste generation 2,633 tons
 - Heat to boil 4 Olympic-sized swimming pools.





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- ICRI
 - 510.1-2013 Electrochemical Techniques to Mitigate the Corrosion of Steel
- NACE
 - SP 0390 Maintenance and Rehabilitation Considerations for Corrosion Control of Atmospherically Exposed Existing Steel-Reinforced Concrete Structures
 - SP 0107 Electrochemical Realkalization and Chloride Extraction for Reinforced Concrete
- FHWA
 - Bridge Preservation Guide
 - Long-Term Effects of Electrochemical Chloride Extraction on Laboratory Specimens and Concrete Bridge Components FHWA-HRT-10-069
 - Several other SHRP studies on ECE and Realkalization
- ACI
 - 222.3R-03 Design and Construction Practices to Mitigate Corrosion of Reinforcement in Concrete Structures







Thank you. Questions?







